

# C.U.SHAH UNIVERSITY

## Summer Examination-2016

Subject Name : Chemistry-VI

Subject Code : 4SC03CHE2

Branch: B.Sc. (Chemistry)

Semester : 3

Date : 28/04/2016

Time : 02:30 To 05:30

Marks : 70

Instructions:

- (1) Use of Programmable calculator & any other electronic instrument is prohibited.
  - (2) Instructions written on main answer book are strictly to be obeyed.
  - (3) Draw neat diagrams and figures (if necessary) at right places.
  - (4) Assume suitable data if needed.
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- Q-1 Attempt the following questions: (14)**
- Arrange the following radiations in their increasing order of energy: Visible, IR, Microwave, Radio. (1)
  - What is the application of electron energy loss spectroscopy? (1)
  - What are the different types of molecular rotors based on symmetry of their structure? (1)
  - What is Beer-Lambert law? (1)
  - Why do we get band instead of line spectra in IR spectroscopy? (1)
  - What is the primary application of IR spectroscopy? (1)
  - What is K-band in UV-Visible spectroscopy? (1)
  - What is the application of Fourier Transform calculations on the time domain spectra of any molecules? (1)
  - What is the characteristic absorption band in the IR Spectra of organic compound containing hydroxyl and nitrile group? (1)
  - What is quenching of fluorescence? (1)
  - Write the mathematical expression of fundamental IR equation. (1)
  - What is static quenching in fluorescence spectroscopy? (1)
  - Define singlet ground state. (1)
  - What is the basic difference between doublet state and triplet state? (1)

**Attempt any four questions from Q-2 to Q-8**

- Q-2 Attempt all questions (14)**
- Explain the principle of spectroscopy associated with transition of atoms or molecules between ground and excited states. (7)
  - What is spectroscopy? Explain different criteria for classification of spectroscopy? (7)
- Q-3 Attempt all questions (14)**
- Write a note on rotational and vibrational spectroscopy. (7)



- b) What is Raman effect? Explain different process involved during collision between a vibrating molecule or lattice and an incident photon. (7)
- Q-4**      **Attempt all questions**      (14)
- a) Discuss different parameters responsible for shifting of the band in UV-Visible Spectroscopy. (7)
- b) Write a note on principle of UV-Visible Spectroscopy. Explain the changes observed in the electronic energy levels, when an unsaturated organic molecule is irradiated with UV-Visible radiation? (7)
- Q-5**      **Attempt all questions**      (14)
- a) Explain the instrumental setup of UV-Visible Spectroscopy. (7)
- b) What are the differences between IR and Raman spectroscopy? (7)
- Q-6**      **Attempt all questions**      (14)
- a) What is Fourier Transform technique in Infrared spectrometer? Draw and explain suitable block diagram. (7)
- b) Why do we observe less number of bands in the IR spectra of CO<sub>2</sub>? Explain the effect of coupled interaction and Fermi resonance in the IR spectra of CO<sub>2</sub>. (7)
- Q-7**      **Attempt all questions**      (14)
- a) Explain the effect of H-bonding in the IR spectra of organic molecules containing hydroxyl group. (7)
- b) What is fluorescence? Draw and explain the Jablonski diagram. (7)
- Q-8**      **Attempt all questions**      (14)
- a) Write a note on resonance energy transfer, fluorescence lifetime and quantum yields. (7)
- b) Explain different types of fluorescence and quenching of fluorescence. (7)

